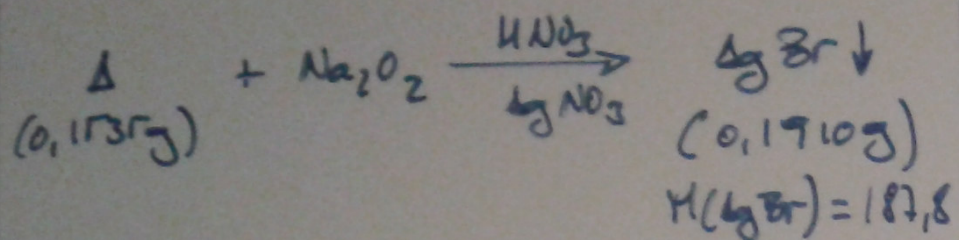


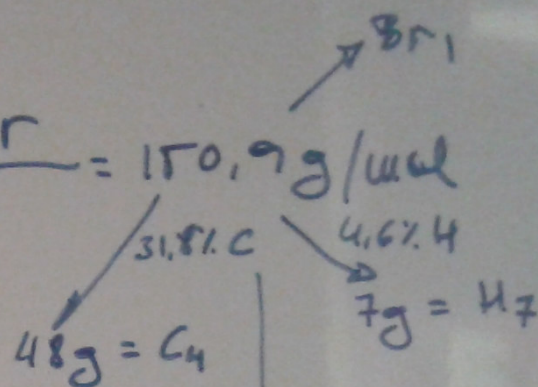
xxxxx budaluciu 23

31,8% C | Δ
4,64% H



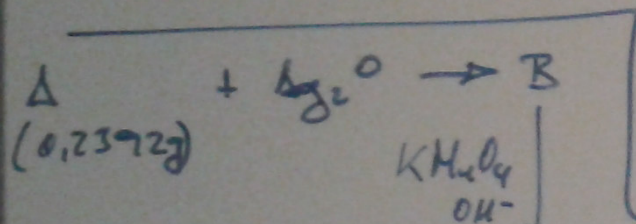
$$\frac{0,1910}{187,8} = \frac{0,1535}{M(\Delta)} \Rightarrow M(\Delta) = 150,9 \text{ g/mol}$$

Si Δ = RBr



Si Δ = RBr₂ $\frac{0,1910}{187,8} = 2 \cdot \frac{0,1535}{M(\Delta)} \Rightarrow M(\Delta) = 302$ (14g)
31,8% C → (96) = C₈
H₁₄

150,9 - 48 - 7 - 77,9 = 16
O₁

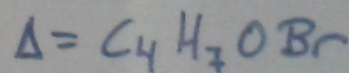
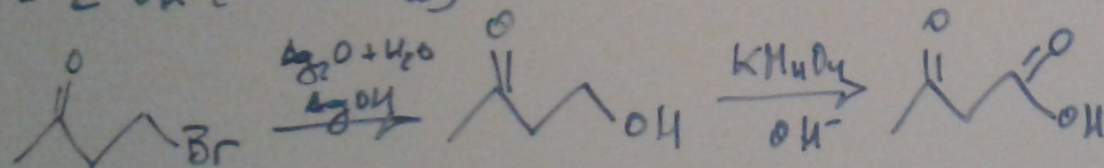


302 - 96 - 14 - 160 = 32 = O₂

CH ≡ 15,84 ml · 0,1 = 1,584 ml

$\frac{239,2 \text{ mg}}{1,584 \text{ ml}} = 151 \text{ g/mol} = M(\Delta)$

B = 2-OH (Primario)



Δ = No es aldehyde (Ag₂O → R-C(=O)OH)
Ni R-C(=O)Br

31,8% C
4,64% H

Δ

xxxx dudalucin 83

C_4H_7OBr

